Stoichiometry Using Molarity Worksheet

For the questions on this worksheet, consider the following equation:

$$Ca(OH)_{2(s)} + 2 HCI_{(aq)} \rightarrow CaCI_{2(aq)} + 2 H_2O_{(l)}$$

- What type of chemical reaction is taking place? _____
- 2) How many liters of 0.100 M HCl would be required to react completely with 5.00 grams of calcium hydroxide?

3) If I combined 15.0 grams of calcium hydroxide with 75.0 mL of 0.500 M HCl, how many grams of calcium chloride would be formed?

- 4) What is the limiting reagent from the reaction in problem #3? _____
- 5) How many grams of the excess reagent will be left over after the reaction in problem 3 is complete?