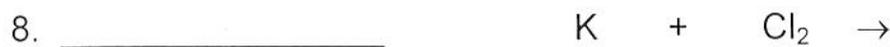
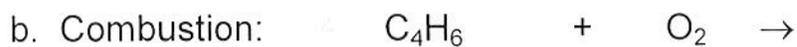


## Predicting Products of Chemical Reactions



9. \_\_\_\_\_  $\text{BaCl}_2 + \text{K}_3\text{PO}_4 \rightarrow$
10. \_\_\_\_\_  $\text{H}_2\text{SO}_4 + \text{KOH} \rightarrow$
11. \_\_\_\_\_  $\text{Al}_2(\text{CO}_3)_3 + \text{heat} \rightarrow$
12. \_\_\_\_\_  $\text{Al} + \text{O}_2 \rightarrow$
13. \_\_\_\_\_  $\text{Pb}(\text{NO}_3)_2 + \text{KOH} \rightarrow$
14. \_\_\_\_\_  $\text{H}_2\text{SO}_4 + \text{BaCl}_2 \rightarrow$
15. \_\_\_\_\_  $\text{Ca} + \text{AgCl} \rightarrow$
16. \_\_\_\_\_  $\text{H}_3\text{PO}_4 + \text{FeBr}_3 \rightarrow$
17. \_\_\_\_\_  $\text{Li} + \text{N}_2 \rightarrow$
18. \_\_\_\_\_  $\text{HCl} + \text{Mg}(\text{OH})_2 \rightarrow$
19. \_\_\_\_\_  $\text{Mg}(\text{OH})_2 + \text{heat} \rightarrow$
20. \_\_\_\_\_  $\text{Fe}(\text{OH})_3 + \text{heat} \rightarrow$