**Molarity Lab**

1) Compare your molarity with the teams on either side of you.

What was the point of using different volumes? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

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2) It is 2018!! There is a search engine called Google. It is your friend.

Google "what is the BCE for calcium carbonate reacting with hydrochloric acid"--it is like magic or maybe even technology!!

You **SAW** bubbling--bubbling means gas production--some of you did not indicate the presence of a gas--do not do that ever again!!

If a physical change or a chemical reaction takes place then give the **BCE** even if I don't ask for it.

**DO NOT** put charges on covalent compounds!!

3) Give me a **data table** EACH and EVERY time--I was actually nice this time--do not test me again!

4) When showing your calculations use English words as well as numbers--tell me what you are doing the calculating!

Mass of beaker and chips - Mass of beaker =

Always identify what the numbers mean--e.g not just mass but mass of chips and beaker before.

6) 40.0 mL of 1 mol/L HCl **IS NOT EQUAL TO** 40.0 g of HCl! You would have died!!!!!