**Making a solution from scratch:**

Make a 15 g/L solution of KoolAid using the volumetric given.

**Calculation:**

**Point form method you used:**

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**Side Question**—the molar mass of KoolAid happens to be 85 g/L.

What is the molar concentration of your solution??

**Calculation:**

**Making a Solution by Dilution**

Using a 100 mL volumetric change your original stock solution to a concentration of 0.080 g/L by dilution.

**Calculation:**

**Labeled Sketch of Your Procedure:**

**Point form method you used:**

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